

Scheme of teaching and examination for 4-Year Part Time Diploma
Ist Semester Diploma in Civil Engineering
THEORY

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Terminal Exam. (A) Marks	Final Exam (B) Marks	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
01	Engineering Mathematics-I	P01101	06	60	03	20	80	100	26	36
02	Engineering Chemistry	P01103	04	50	03	20	80	100	26	36
03	Language & Communication Skill	P01105	04	60	03	20	80	100	26	36
Total period per week :-			14		Total :-			300		

PRACTICAL

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Marks Internal Exam. (A)	Marks External Exam. (B)	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
04	Engineering Chemistry Lab.	P01107	04	50	03	10	40	50	16	21
05	Workshop Practice	P01108	12	120	06	10	40	50	16	21
Total period per week :-			16		Total :-			100		

SESSIONAL

Sl. No	Subjects	Subject Code	Teaching Scheme		Examination Scheme			
			Periods per week	Period in one session (year)	Marks of Internal Examiner (X)	Marks of External Examiner (Y)	Total Marks (X+Y)	Pass marks in the subjects
06	Workshop Practice	P01109	-	-	40	60	100	50
07	Student Centered Activity	P01209	04	50	20	30	50	25
Total period per week:-			04		Total :-		150	
Over all total period per week:-			34		Total Marks:-		550	

Scheme of teaching and examination for 4-Year Part Time Diploma
Ist Semester Diploma in Electrical Engineering
THEORY

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Terminal Exam. (A) Marks	Final Exam (B) Marks	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
01	Engineering Mathematics-I	P01101	06	60	03	20	80	100	26	36
02	Engineering Chemistry	P01103	04	50	03	20	80	100	26	36
03	Language & Communication Skill	P01105	04	60	03	20	80	100	26	36
Total period per week :-			14		Total :-			300		

PRACTICAL

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Marks Internal Exam. (A)	Marks External Exam. (B)	Total Marks (A+B)	Pass marks final exam	Pass marks in the subjects
04	Engineering Chemistry Lab.	P01107	04	50	03	10	40	50	16	21
05	Workshop Practice	P01108	12	120	06	10	40	50	16	21
Total period per week :-			16		Total :-			100		

SESSIONAL

Sl. No	Subjects	Subject Code	Teaching Scheme		Examination Scheme				
			Periods per week	Period in one session (year)	Marks of Internal Examiner (X)	Marks of External Examiner (Y)	Total Marks (X+Y)	Pass marks in the subjects	
06	Workshop Practice	P01109	-	-	40	60	100	50	
07	Student Centered Activity	P01209	04	50	20	30	50	25	
Total period per week:-			04		Total :-			150	
Over all total period per week:-			34		Total Marks:-			550	

Scheme of teaching and examination for 4-Year Part Time Diploma
1st Semester Diploma in Electronics Engineering
THEORY

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Terminal Exam. (A) Marks	Final Exam (B) Marks	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
01	Engineering Mathematics-I	P01101	06	60	03	20	80	100	26	36
02	Engineering Chemistry	P01103	04	50	03	20	80	100	26	36
03	Language & Communication Skill	P01105	04	60	03	20	80	100	26	36
Total period per week :-			14		Total :-			300		

PRACTICAL

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Marks Internal Exam. (A)	Marks External Exam. (B)	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
04	Engineering Chemistry Lab.	P01107	04	50	03	10	40	50	16	21
05	Workshop Practice	P01108	12	120	06	10	40	50	16	21
Total period per week :-			16		Total :-			100		

SESSIONAL

Sl. No	Subjects	Subject Code	Teaching Scheme		Examination Scheme				
			Periods per week	Period in one session (year)	Marks of Internal Examiner (X)	Marks of External Examiner (Y)	Total Marks (X+Y)	Pass marks in the subjects	
06	Workshop Practice	P01109	-	-	40	60	100	50	
07	Student Centered Activity	P01209	04	50	20	30	50	25	
Total period per week:-			04		Total :-			150	
Over all total period per week:-			34		Total Marks:-			550	

Scheme of teaching and examination for 4-Year Part Time Diploma
1st Semester Diploma in Mechanical Engineering
THEORY

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Terminal Exam. (A) Marks	Final Exam (B) Marks	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
01	Engineering Mathematics-I	P01101	06	60	03	20	80	100	26	36
02	Engineering Chemistry	P01103	04	50	03	20	80	100	26	36
03	Language & Communication Skill	P01105	04	60	03	20	80	100	26	36
Total period per week :-			14		Total :-			300		

PRACTICAL

Sl. No	Subjects	Subject code	Teaching Scheme		Examination Scheme					
			Periods per week	Period in one session (Year)	Hours of Exam.	Marks Internal Exam. (A)	Marks External Exam. (B)	Total Marks (A+B)	Pass marks in final exam	Pass marks in the subjects
04	Engineering Chemistry Lab.	P01107	04	50	03	10	40	50	16	21
05	Workshop Practice	P01108	12	120	06	10	40	50	16	21
Total period per week :-			16		Total :-			100		

SESSIONAL

Sl. No	Subjects	Subject Code	Teaching Scheme		Examination Scheme			
			Periods per week	Period in one session (year)	Marks of Internal Examiner (X)	Marks of External Examiner (Y)	Total Marks (X+Y)	Pass marks in the subjects
06	Workshop Practice	P01109	-	-	40	60	100	50
07	Student Centered Activity	P01209	04	50	20	30	50	25
Total period per week:-			04		Total :-			150
Over all total period per week:-			34		Total Marks:-			550

ENGINEERING MATHEMATICS-I

Subject Code P01101	Theory			No of Period in one session : 60		
	No. of Periods Per Week			Full Marks	:	100
	L	T	P/S	Annual Exam.	:	80
	06	—	—	Internal Exam.	:	20

Rationale:

The Subject Engineering Mathematics is being introduced into the Diploma Course to provide mathematical background to the students so that they can be able to grasp the engineering subjects properly. This course will enable them to analyse and understand the engineering problems scientifically based on Mathematics.

The subject is divided into two papers, viz. Engineering Mathematics - I and Engineering Mathematics - II. The paper Engineering Mathematics - I consists of the following:

1. Algebra
2. Trigonometry
3. Co-ordinate Geometry

The details are given in the curriculum:

Objectives:

- By covering the course in Engineering Mathematics - I, the students will be able to:
- Know Sequence & Series, Permutations and Combinations, Binomial Theorem, Determinates and Matrices, Properties of Triangles, Solution of Trigonometrical equations, Inverse Circular functions, complex quantities, co-ordinate systems, equations of lines, circles, equations of lines in three dimensions, equation of plane,
 - Understand their engineering applications.
 - Solve related simple numerical problems which will enable them to understand the subject.

S.No.	Topics	Periods
01	Algebra - Sequence & Series - Principle of Mathematical Induction - Permutation and Combination - Binomial Theorem - Determinants and Their Properties - Matrix Algebra - Complex Number	(30)
02	Trigonometry - Trigonometrical ratios of compound angles up to conditional Trigo nometrical Identities - Properties of Triangle - Logarithm - Solution of Triangles & General Value - Inverse Circular Function	(15)
03	Co-ordinate Geometry - Two dimensional : upto equation of circles - Three dimensional: upto straight line	(15)

CONTENTS:

TOPIC: 01 - ALGEBRA:

		Periods
01.01	Sequence & Series: Arithmetic Progression (A.P.), Simple examples of A.P., Geometrical Progression (G.P.), Sum to infinity of a G.P., Sum of Squares and cubes of a naturals, idea of Harmonic Progression (H.P.), Relation between Arithmetic mean, Geometrical Mean and Harmonic mean. Insertions of AMs, GMs & HMs between two numbers.	[08]
01.02	Principle of Mathematical Induction	[02]
01.03	Permutations & Combinations: Introduction, Fundamental Principle of counting; The Factorial; Permutations, Simple practical problems on permutation; Combinations; simple practical problems on combinations.	[04]
01.04	Binomial Theorem: Binomial Theorem for positive Index, Some applications of Binomial Theorem for any Index, Idea of Exponential and Logarithmic Series. (Simple Problem).	[04]

01.05	Determinates: Determinants and their fundamental properties, simple problem, Difference between determinant and a matrix.	[02]
01.06	Matrices: - Different types of Matrices - Algebra of Matrices - Transpose, Adjoint & Inverse of Matrices - Solution of linear simultaneous equations by matrix method	[04]
01.07	Complex Numbers: Idea of a complex number, its geometrical representation, Modulus and Amplitude, Conjugate of a Complex number, Addition & Subtraction of a complex number with geometric notation, Multiplication and Division of one complex number by another with geometric representation. Idea of DeMoivre's Theorem, Roots of a Complex and Cube root of unity.	[04]
01.08	Number System: Binary, octal, Decimal & Hexadecimal system. Radix conversion. Idea of Boolean Algebra	[02]

TOPIC: 02 - TRIGONOMETRY:

02.01	Trigonometrical ratios of Compound angles. Trigonometrical ratios of Multiple sub-multiple angles, transformation formulae & conditional Trigonometrical identities.	[04]
02.02	Properties of Triangle: Relations between the side and angles of a triangle. Simple problems based on it.	[04]
02.03	Logarithm: Definition, Fundamental Rules and properties of Logarithms.	[02]
02.04	General Values and Inverse Functions: Formulae for all angles which have a given Sine, Cosine and Tangent. Formulae for angles both equi-sinal and equi-cosinal Inverse Circular Functions, Solution of Equations expressed in inverse notation.	[05]

TOPIC: 03 - CO-ORDINATE GEOMETRY:

03.01	Two Dimensional Co-ordinate Geometry	
03.01.01	Idea of cartesian and polar co-ordinate systems. Relations between them.	[01]
03.01.02	Distance between two points, section formula and Area of Triangle. Intelligent questions based on these (cartesian system only), centroid and incentre of a triangle.	[02]
03.01.03	Equations of Locus: Equation of a straight line in different forms. Angle between two straight lines and their deduction, equation of circle, simple problem.	[04]
03.02	Three Dimensional Co-ordinate Geometry	
03.02.01	Co-ordinates of a point, Distance between two points, Section formula (Cartesian system only)	[01]
03.02.02	Direction Cosines, Angle between two lines, Important deductions.	[02]
03.02.03	Plane, Projection of the join of two points on a plane, Equation of plane, Angle between two planes, Important deductions.	[02]
03.02.04	Equation of a straight line as intersection of two planes, Symmetric form of a straight line, simple problem.	[03]

Books Recommended:

Engineering Mathematics - I

1.	Mathematics for Class XI Part I	-	NCERT/R. S. Aggawal/R.D.Sharma
2.	Mathematics for Class XI Part II	-	NCERT/R. S. Aggawal/R.D.Sharma
3.	Mathematics for Class XII Part I	-	NCERT/R. S. Aggawal/R.D.Sharma
4.	Mathematics for Class XII Part II	-	NCERT/R. S. Aggawal/R.D.Sharma
5.	Algebra		Dr. K.C. Sinha/ Lalgi Pd./Das & Gupta
6.	Trigonometry		Dr. K.C. Sinha/ Lalgi Pd./Das & Gupta
7.	Co-ordinate geometry		Dr. K.C. Sinha/ Lalgi Pd./Das & Gupta
8.	Solid geometry		Dr. K.C. Sinha/ Lalgi Pd./Das & Gupta

Reference Books:

1.	Engineering Mathematics - Part I & Part II	-	H.K. Dass, S. Chand & Co.
2.	Polytechnic Mathematics for Diploma level	-	H.K. Dass, S. Chand & Co.

ENGINEERING CHEMISTRY

Subject Code P01103	Theory			No of Period in one session : 50		
	No. of Periods Per Week			Full Marks	:	100
	L	T	P/S	Annual Exam.	:	80
	04	—	—	Internal Exam.	:	20

Rationale & Objective:

Keeping in view the recent developments in Science and the present needs in Industries, the curriculum of Engineering Chemistry has been revised so that the Engineers or Technicians may have a better knowledge of Chemistry, especially regarding the application of the subject in various fields of Industries. An emphasis, in this direction, has been made in the curriculum.

A new chapter on Environmental Chemistry has been introduced to make the students acquainted with the various pollution hazards which is becoming more critical everyday.

The following topics are so chosen that through their contents the students are able to develop knowledge, skill and scientific attitude. It will enable them to distinguish, differentiate, analyse and solve engineering problems.

S.No.	Topics	Periods
GROUP - A		
1.	Importance of Chemistry for Engineers and its applications in industries	(02)
2.	General Chemistry	(05)
3.	Atomic Structure	(03)
4.	Chemical Bonding	(03)
5.	Chemical Equilibrium	(03)
6.	Metallurgical Operations	(08)
1.	Water Treatment	(08)
2.	Fuel & Combustion	(08)
3.	Lubricants	(02)
4.	Paints and Varnishes	(02)
5.	Environmental Chemistry	(06)

CONTENTS:

GROUP - A

	Topic: 01 - Introduction	[02]
01.01	Importance of Chemistry for Engineers and its application in industries.	
	Topic: 02 - General Chemistry	[05]
02.01	Atomic Wt. Equivalent Wt., Molecular Wt. and their determination, Numerical Problems.	
02.02	Mole Concept, Avogadro's number, Numerical Problems.	
	Topic: 03 - Atomic Structure	[03]
03.01	Basic idea of fundamental particles, Atomic Number, Mass Number, Rutherford model & Bohr's model.	
03.02	Electronic configuration in s, p, d, f notation.	
	Topic: 04 - Chemical Bonding	[03]
04.01	Ionization Potential, Electron affinity, electronegativity.	
04.02	Types of Chemical Bonds - Electrovalent, Covalent (Polar and non-polar) and Co-ordinate bonds.	
	Topic: 05 - Chemical Equilibrium	[03]
05.01	Reversible and Irreversible reaction, Chemical Equilibrium.	
05.02	Law of mass action.	
05.03	Ionic product of water, PH-scale, Common Ion Effect and Numerical problems.	
	Topic: 06 - Metallurgical Operations	[08]
06.01	General metallurgical operations, Concentration of metal ore, Roasting, Calcination, Smelting, refining of metals.	
06.02	Extraction of Iron, Aluminium and Copper.	
06.03	Manufacture of steel - (a) Bessemer process, (b) Open Hearth process, effect of impurities such as Mn, P, S and Si. Heat treatment of steel, Annealing, Hardening, Tempering, Normalising, Case hardening, Nitriding and Cyaniding	
06.04	Introduction, Importance, Classification and uses of alloys with examples.	

GROUP - B

Topic: 07 - Water Treatment

[08]

- 07.01 Introduction - Use of water for Industrial and domestic purposes, sources of water supply.
07.02 Hardness of water, degree of hardness and its estimation (Hehner and EDTA methods). Numerical problems on degree of hardness. PH-value of water, disinfection of water and Municipal Supply.
07.03 Softening of hard water (Lime-Soda method, Permutit, Ion Exchange and calgon methods).

Topic: 08 - Fuel and Combustion

[08]

- 08.01 Introduction - Importance of fuels in Industries, classification of fuels, calorific values, Determination of calorific value and Numerical problems. Characteristics of an ideal fuel.
08.02 Refining and cracking of petroleum, knocking. Octane Number and Cetane Number. Merits and demerits of fuels, L.P.G., Coal gas, Oil gas and Producer gas.

Topic: 09 - Lubricants

[02]

- 10.01 Introduction & Classification of lubricants.
10.02 Properties of lubricants, Lubricants Oil, grease, emulsions.

Topic: 10 - Paints and Varnishes

[02]

- 11.01 Characteristics of a good paint, brief study of various constituents of a paint.

Topic: 11 - Environmental Chemistry

[06]

- 13.01 Introduction:
13.01.01 Effect of pollution on human health (Name of diseases) and plant.
13.02 Air Pollution:
13.02.01 Causes of air pollution like factory Smoke discharge, Automobile exhaust gas, Deforestation etc.
13.02.02 Brief idea of pollution effects like Acid rain, Green house effect, Action of Ozone layer which causes green house effect on earth, effect of chlorofluorocarbon on depletion of ozone layer.
13.03 Water Pollution:
13.03.01 Standard prescribed by WHO, IMC and Bureau of Indian Standard for pure drinking water.

Books Recommended:

- | | | | |
|---|---------------------------------------|---|-------------------------------|
| 1 | Text Book of Engineering Chemistry | - | M.M. Uppal |
| 2 | Text Book of Engineering Chemistry | - | C.V. Agrawal |
| 3 | Text Book of Engineering Chemistry | - | P.C. Jain |
| 4 | Pradyogiki Rasayan (Hindi) | - | S.Z. Aahmad & Prof. Subuktgin |
| 5 | Takniki Rasayan Bhag 1 evam 2 (Hindi) | - | Roop Prakashan |
| 6 | a. Inorganic Chemistry | - | P.L. Soni |
| | b. Physical Chemistry | - | P.L. Soni |
| 7 | a. Inorganic Chemistry | - | Biltu Singh |
| | b. Physical Chemistry | - | Biltu Singh |
| 8 | a. Inorganic Chemistry | - | Ram Ratan Pd. |
| 9 | Environmental Chemistry | - | |

LANGUAGE & COMMUNICATION SKILL (ENGLISH & HINDI)

Subject Code P01105	Theory			No of Period in one session : 60		
	No. of Periods Per Week			Full Marks	:	100
	L	T	P/S	Annual Exam.	:	80
	04	-	-	Internal Exam.	:	20

Rationale & Objective:

The primary aim of this course is to help technical students studying in Polytechnics and Engineering Institutes acquire the skills of language and communication in order to be successful in their studies and subsequent professional life. It has been found that in the world of work of diploma holder they have to perform various job functions like Letter Writing, maintaining office records, drawing up tender notices, writing technical reports, communicating with subordinate staff and/or labourer and with superiors.

The curriculum has been designed to improve the knowledge of the Language, comprehension and its application to develop communication skill.

The curriculum also seeks to develop the student's power of oral communication through effective use of body language and necessarily puts knowledge to practice through exposure in varied form.

The curriculum has been designed both in English & Hindi languages.

S.No.	Group	Topic	Periods	Marks
1	A	ENGLISH	30	50
2	B	HINDI	30	50
Total:			60	100

GROUP - A [ENGLISH]

S.No.	Topic	Periods
01	A. Language Practice	[08]
	B. Oral Communication	[05]
02	Comprehension	[03]
03	Paragraph Writing	[02]
04	Letter Writing	[04]
05	Tender Notice & Advertisement	[04]
06	Report Writing	[04]
Total:		30

CONTENTS:

TOPIC 01(A) - Language Practice: [08]

- 01.01 Tenses of verbs
- 01.01.01 Writing about the Present
 - Subject verb agreement
 - Negative statements
 - Is/ Are VERB – ed (is needed, are powered etc.)
- 01.01.02 Writing about the Past
 - VERB - ed (Past Simple)
 - Was/ Were VERB - ed (Past simple passive)
 - Has/ Have VERB - ed
 - Has/ Have been VERB - ed
 - Has VERB - ed + VERB - ed (Past perfect + Past simple)

(The demonstration has already started before the office broke for lunch)

Was/ Were VERB - ed + VERB – ed (Past continuous + past simple)

- 01.01.03 Writing about the Future
- Shall/ Will VERB (Future simple)
 - Shall/ Will be VERB - ed (Future simple passive)
- 01.02 Auxiliaries
- Use of can could, will would, shall should, may might etc.
(Drilling exercise with suitable examples to be done)
- 01.03 Word Formation
- Common roots in Technical English
 - Noun endings, -tion, -ment, -ance, -ity, -logy, -meter, -metry, -or, -er etc.
 - Prefixes that mean NOT: in, on, non, il, im, de, dis, mis, mal
 - Words that end with: -ize, -ate, -ify
 - Adjectives that end with: -al, -ic, -ical, -ar, -ary, -ory, -ing
- 01.04 Single Word Substitution
- Drilling of exercise
(Page No. 147 to 151 of Text Book)
- 01.05 Sentence Structure
- Completing, joining, reframing (for emphasis) and transformation of sentences
- 01.06 Punctuation
- Correct use of comma, semi-colon, colon, full stop, apostrophe, inverted commas, note of exclamation, note of interruption, dash, brackets, hyphen, capital letters and italics.

TOPIC 01(B) - Oral Communication:

[05]

- Manners & basic etiquettes
Body Language - the role of body postures, movements, gestures, facial expressions, dress & make up in effective communication
- Information/ Desk/ Front Office/ Telephone conversation
(Practice with audio/ video cassettes)
- Conduct while facing interviews (Mock Interview)
- Group discussions, debates, elocution

TOPIC 02 - Comprehension:

[03]

- 02.01 Prescribed unit from communication in English for Technical Students (Orient Longman):
- i. Uses of Mango Wastes
 - ii. Making Money in India
 - iii. Radar: its operation and benefits
 - iv. Technology for Mankind

TOPIC 03 - Paragraph Writing:

[02]

- 03.01 General – Specific
- 03.02 Process – Description
- 03.03 Problem – Solution
- 03.04 Data – Comment

TOPIC 04 – Letter/ Application Writing: [04]

04.01 Official letters to an from higher authorities/ departments regarding administrative/ establishment/ financial matters.

04.02 Commercial letters regarding enquiries/ proposals for purchase/ service.

04.03 Drafting application for jobs - format, style & contents

TOPIC 05 - Tender Notice & Advertisement: [04]

05.01 Inviting Tenders/ Quotations - format & contents, formalities involved, placing orders.

05.02 Drafting advertisements for 'situation vacant'/ 'situation wanted' columns, for sale/ purchase of items etc.

TOPIC 06 - Report Writing: [04]

06.01 Types, structure and utility of reports

06.02 Technical reports

06.02.01 Project reports

06.02.02 Enquiry reports

06.02.03 Stock verification reports etc.

(The teacher should help the students in the preparation of their project report)

Books Recommended:

Text Book:

1. Communication in English for Technical Students - prepared by C.D.C., T.T.T.I. Calcutta (Orient Longman)

Reference Books:

1. An Intermediate English Practice Book - by S. Pit Corder (Orient Longman)
2. Living English Structure - by W.S. Allen (Orient Longman)
3. Advance Learner's Dictionary - by A.S. Hornby (O.U.P.)

GROUP - B [HINDI]

क्रम सं.	पाठ्य	व्याख्यान
01	भाषा अभ्यास	[08]
02	मौखिक सम्प्रेषण	[05]
03	अपठित गद्यांश और प्रश्नोत्तर	[03]
04	अनुच्छेद लेखन	[02]
05	पत्र/ आवेदन लेखन	[04]
06	निविदा सूचना एवं विज्ञापन	[04]
07	प्रतिवेदन लेखन	[04]
कुल:		[30]

CONTENTS:

पाठ्य 01 – भाषा अभ्यास: [08]

01-01 शब्द रचना

01-01-01 विशेषण

विशेष्य और विशेषण की रचनाएँ विशेषण बनाने के कुछ नियम, पद वाचक विशेषण

01-01-02 मूल शब्द, उपसर्ग, प्रत्यय

01-01-03 विदेशी शब्दों का हिन्दी प्रयोग

01-02 वाक्य रचना

01-02-01 वाक्य का रूपान्तर, सामान्य वाक्य, वाक्य उपवाक्य, वाक्य की अशुद्धियाँ, वाक्य में कर्ता और क्रिया का मेल, संज्ञा और सर्वनाम का मेल, वाक्यज्ञत प्रयोग।

01-03 विराम चिन्ह

01-04 . विपरीतार्थक शब्द

. युग्म शब्द

. अनेक शब्दों के लिए एक शब्द

. एक शब्द और विभिन्न प्रयोग

. एक शब्द का विभिन्न शब्द भेदों में प्रयोग

. संक्षेपण

पाठ्य 02 – मौखिक सम्प्रेषण: [05]

02-01 तौर तरीके एवं आधारभूत शिष्टाचार

02-02 शारीरिक भाषा – शारीरिक भावभंगिमा द्वारा सम्प्रेषण, अतिविहित, संकित मुखाकृति द्वारा सम्प्रेषण, पोशाक तथा प्रसाधन द्वारा प्रभावकारी सम्प्रेषण

02-03 जानकारी/ डेस्क/ कार्यालय का अग्रभाग/ टेलीफोन वार्तालाप (श्रष्टा/ दृश्य कैसेटों द्वारा अभ्यास)

02-04 अन्तर्वीक्षा के समय आचरण

02-05 सामूहिक परिचर्चा, वाद-विवाद, वक्तृता

पाठ्य 03 – अपठित गद्यांश और प्रश्नोत्तर: [03]

सम्बद्ध पाठ्यक्रम समसामयिक पत्रिका, अखवार एवं पुस्तक में सम्पादकीय तथा लेख पर आधारित होंगे। परीक्षा अपठित अवतरणों पर आधारित होगी, शब्दार्थ, तर्क, विचार, वाक्य संरचना, वाक्य संरचना एवं प्रयोग के सम्बन्ध में विशेषकर वस्तुनिष्ठ प्रश्न पूछे जायेंगे।

- पाठ्य 04 – अनुच्छेद लेखन:** [05]
- 04-01 सामान्य – विशेष
- 04-02 प्रक्रिया – वर्णन
- 04-03 समस्या – समाधान
- 04-04 अकिंछा – समीक्षा
- पाठ्य 05 – पत्र/ आवेदन लेखन:** [04]
- 05-01 उच्चाधिकारियों/ विभागों के साथ प्रशासनिक/ स्थापना/ वित्तीय मामलों से सम्बन्धित पत्राचार।
- 05-02 पूछताछ/ क्रय/ सेवा से सम्बन्धित पत्राचार।
- 05-03 नियोजन हेतु आवेदन
- पाठ्य 06 – निविदा सूचना एवं विज्ञापन:** [04]
- 06-01 निविदा/ कोटेशन आमंत्रित करना – रूपरेखा एवं संदर्भ सम्बद्ध औपचारिकता, आदेश।
- 06-02 रिक्तियाँ/ आवश्यकता/ क्रय/ विक्रय आदि के लिये विज्ञापन का प्रारूप।
- पाठ्य 07 – प्रतिवेदन लेखन:** [04]
- 07-01 प्रतिवेदन के प्रकार, संरचना एवं उपयोगिता।
- 07-02 तकनीकी प्रतिवेदन – परियोजना प्रतिवेदन, जाँच प्रतिवेदन आदि (परियोजना प्रतिवेदन तैयार करने में शिक्षक को विद्यार्थियों की मदद करनी चाहिए)

निर्धारित पुस्तकें

टेक्स्ट बुक(पाठ्य पुस्तक)/ संदिर्ग पुस्तकें

- | | | |
|--------------------------------------|---|--|
| 1. आधुनिक हिन्दी व्याकरण और रचना | - | डा. वासुदेव नन्दन प्रसाद, भारती भवन, पटना |
| 2. हिन्दी में उन्नत टिप्पण और सार | - | राम विनायक सिंह, लोक भारती प्रकाशन, इलाहाबाद |
| 3. हिन्दी में प्रशासनिक पत्र लेखन | - | राम विनायक सिंह, लोक भारती प्रकाशन, इलाहाबाद |
| 4. हिन्दी प्रारूपण और टिप्पण | - | मल्होत्रा, फ्रेजर रोड, पटना |
| 5. शिक्षार्थी हिन्दी शब्दकोश | - | डा. हरदेव वाहरी, रामपाल एण्ड सन्स |
| 6. अंग्रेजी हिन्दी शासकीय प्रयोग कोश | - | गोपीनाथ श्रीवास्तव, सम पाल एण्ड सन्स |

ENGINEERING CHEMISTRY (LAB)

Subject Code P01107	Practical			No of Period in one session : 50		
	No. of Periods Per Week			Full Marks	:	50
	L	T	P/S	Annual Exam.	:	40
	—	—	04	Internal Exam.	:	10

Rationale & Objective:

The Chemistry Lab. Practical has been introduced with a view to develop scientific attitude among the students. The topics (experiments) have been chosen to develop skill among the students so that they can measure, differentiate and analyse the best results. This will help them solve the engineering problems in their world of work.

S.No.	Topics	Periods
(At least ten experiments are to be performed)		
1	Preparation of derivatives	
2	Titration	
3	Quantitative Analysis	
4	Quantitative Analysis of Simple Inorganic Salts	
5	Qualitative and Quantitative Analysis of drinking water	

CONTENTS:

Topic: 01 - Preparation of derivatives

- 01.01 Preparation of Barium Sulphate from Barium Chloride.
- 01.02 Preparation of Copper Sulphate from Copper Carbonate.
- 01.03 Preparation of Copper Sulphate from Copper Nitrate.
- 01.04 Preparation of Copper Chloride from Copper Sulphate.
- 01.05 Preparation of Calcium Carbonate from Calcium Oxide.

Topic: 02 - Titration

- 02.01 Preparation N/10 solution of oxalic acid and Sodium Carbonate
- 02.02 Standardisation of the given solution of NaOH or KOH with the help of N/10 Oxalic acid solution.
- 02.03 Determination of the volume of a drop of water.
- 02.04 To determine the quantity of Na₂CO₃/litre in a mixture of Na₂CO₃ and NaOH solution.

Topic: 03 - Quantitative Analysis

- 03.01 Determination of percentage of calcium or calcium carbonate in a given sample of calcium carbonate.
- 03.02 Determination of percentage of moisture in a given sample of coal..

Topic: 04 - Qualitative Analysis

- 04.01 Analysis of simple inorganic salts containing not more than two radicals among the following :-
- | | | | | | | | | | |
|------------------------------|---------------------------------|--------------------|--------------------|---------------------|--------------------------------|--------------------------------|--------------------------------|--------------------|---------------------|
| Pb ⁺⁺ , | Hg ⁺⁺ , | Cu ⁺⁺ , | Cd ⁺⁺ , | Bi ⁺⁺⁺ , | As ⁺⁺⁺ , | Sb ⁺⁺⁺ , | Fe ⁺⁺ | or | Fe ⁺⁺⁺ , |
| Al ⁺⁺⁺ , | Cr ⁺⁺⁺ , | Mn ⁺⁺ , | Zn ⁺⁺ , | Co ⁺⁺ , | Ca ⁺⁺ , | Sr ⁺⁺ , | Ba ⁺⁺ , | Mg ⁺⁺ , | Na ⁺ , |
| K ⁺ , | NH ₄ ⁺⁺ , | Cl ⁻ , | Br ⁻ , | I ⁻ , | NO ₃ ⁻ , | CO ₃ ⁻ , | SO ₄ ⁻ , | S ⁻ , | and |
| NO ₂ ⁻ | | | | | | | | | |

Topic: 05 - Qualitative & quantitative Analysis of Drinking Water

Note :- Water samples from five different sources, Well, handpump, water supply etc. from neighbourhood to be collected by each group of two students and following tests to be conducted :-
Qualitative Analysis (with the help of field test kits available) or the following :-

- i. Total Solid dissolved.
- ii. Chlorine.
- iii. Flourine.
- iv. Iron.
- v. Nitrite.
- vi. Nitrate.
- vii. Sulphide/Sulphate.

Quantitative Analysis in the laboratory

- i. pH-Value-By pH meter.
- ii. Chlorine- By Gravimetric method.
- iii. Sulphate- By Gravimetric method.

WORKSHOP PRACTICE

Subject Code P01108	Practical			No of Period in one session : 120		
	No. of Periods Per Week			Full Marks	:	50
	L	T	P/S	Annual Exam.	:	40
	—	—	12	Internal Exam.	:	10

Rationale & Objective:

A Diploma holder technician must know how to work on shop floor. This helps to develop psychomotor skill and attitude. The knowledge & skill to use machines, equipment, tools and measuring instruments is required to be developed. Safe handling of machines and tools is also very important. So, it is essential for students of 1st year to undergo basic workshop practical training. The topics include practical works in carpentry, welding, fitting, smithy sheet metal shop & machine shop. It is required to inculcate safe habits and attitude so that accidents are avoided at every step. Topics have been prescribed to fulfil these objectives.

The students are supposed to come in proper workshop dress. Wearing shoes in the workshop is compulsory.

<u>S.No.</u>	<u>Topic</u>	<u>No. of Jobs</u>	<u>No. of Periods</u>
01	Safety precautions and knowledge of hand tools	--	(03)
02	Duty & Responsibility of staffs working difference section.	02	(03)
03	Wood working (carpentry section)	02	(30)
04	Fitting Section	02	(30)
05	Blacksmithy Section	02	(20)
06	Welding	02	(19)
07	Sheet metal work	02	(15)
			(120)

CONTENTS:

TOPIC: 01 - SAFETY PRECAUTIONS & KNOWLEDGE OF HAND TOOLS: **[03]**

- 01.01 Importance, general safety precautions on different shop floors.
- 01.02 Personal, tools and general safety.

TOPIC:02 DUTY & RESPONSIBILITY OF STAFFS WORKING DIFFERENCE SECTION **(03)**

TOPIC: 03 - WOOD WORKING (CARPENTRY SECTION): **[30]**

- 03.01 Carpentry Practice
- 03.01.01 Use of hand tools for holding drilling, cutting, marking & mixed tools such as vice, clamps, saw, hammers, mallet, screwdriver etc. **[03]**
- 03.01.02 Different carpenter joints & their application (Mortish & Tanon, Dovetail, half lap etc. **[03]**
- 03.02 Identification of joint in a particular job articles of furniture items. **[04]**
- 03.03 Jobs to be made: **[20]**
- 03.03.01 Wall hanger
- 03.03.02 Pulse mixer

TOPIC: 04 - FITTING SECTION: **[30]**

- 04.01 Importance of fitting operation such as chipping, sawing, filling, scraping, drilling, reaming etc. **[03]**
- 04.02 Functions, classification of tools, work holding and clamping specific tools for example File (length, type, grade of cut etc.) vices, cold chisel, hand tools etc. **[05]**
- 04.03 Use of hand dies & tape for pipe work (water and sans) **[03]**
- 04.04 Fitting practice & jobs **[19]**
- 04.04.01 Male female joint - 01
- 04.04.02 Chipping, Filling, Scraping - 01
- 04.04.03 Marking, fitting

TOPIC: 05 - BLACKSMITHY SECTION: [20]

05.01	Introduction to smithy tools and their uses	[03]
05.02	Smithy Practice (forging)	[03]
05.02.01	Smithy operation such as offsetting, drawing, bending, welding round to square section and vice-versa.	
05.03	<u>Jobs to be made:</u>	[14]
05.03.01	Chiesel	
05.03.02	Ring	
05.03.03	Punch	
05.03.04	Screw Driver	

TOPIC: 06 - WELDING: [19]

	Before starting welding, the Foreman/ Instructor should show to the students the methods of line testing, working of iron clad switches, knife switches.	[03]
	By observation a student is able to:	
	- Identify welding materials	
	- Understand difference between gas welding & electric welding	
	- Understand difference between welding & soldering	
	- Know the materials which can be welded and materials which can not be welded.	
06.01	Introduction to gas welding.	[05]
06.02	Use of welding equipment and tools and accessories including Personal Protective requirement such as Boot, Gloves, safety goggles, Apron etc.	
06.03	Welding Practice	[11]
06.03.01	Butt joint	
06.03.02	"T" joint	
06.04	Introduction to brazing process, filler material and fluxes application of brazing.	

TOPIC: 07 - SHEET METAL WORK: [15]

07.01	Introduction to sheet metal, procedure and safety precautions.	[03]
07.02	Aquaintance with sheet metal tools and their safe use.	
07.03	Sheet metal practice.	[03]
07.03.01	Simple Development and cutting, bending and shearing of sheet metal	
07.03.02	Marking	[09]
07.03.03	Filing & Finishing	
07.03.04	Fabrication of a sheet metal:	
	- Cabinet	
	- Conical funnel	

Books Recommended for Workshop Practice (Practical):

- | | |
|---------------------------------|----------------------------------|
| 1. Shop Theory | - By Anderson (Tata McGraw Hill) |
| 2. Workshop and Tools Hand Book | - Audel Series |
| 3. Workshop Technology | - Hajra & Choudhary |

Reference Books:

- | | |
|----------------------|------------------------------|
| 1. Workshop Practice | - Rajeev Upadhayay, |
| 2. Workshop Practice | - by N.T.T.T.I. Chanandigarh |

WORKSHOP PRACTICE

Subject Code P01109	Sessional			No of Period in one session :		
	No. of Periods Per Week			Full Marks	:	100
	L	T	P/S	Annual Exam.	:	60
	—	—	—	Internal Exam.	:	40

<u>S.No.</u>	<u>Topic</u>	<u>No. of Jobs</u>
1.	Wood Work (carpentry section):	
	(a) Wall Hanger	01
	(b) Pulse Mixer	01
2.	Fitting Section:	
	(a) Male-Female joint	01
	(b) Chipping, filing and scraping	01
3.	Blacksmithy Section:	
	(a) Chiesel	01
	(b) Ring	01
4.	Welding Section:	
	(a) Butt joint	01
	(b) "T" joint	01
5.	Sheet Metal Work:	
	(a) Fabrication of a sheet metal cabinet	01
	(b) Conical Funnel	01
		(10) jobs

STUDENT CENTERED ACTIVITIES
(Language & communication skill)
English & Hindi

Subject Code P01209	Sessional			No of Period in one session : 50		
	No. of Periods Per Week			Full Marks	:	50
	L	T	P/S	Annual Exam.	:	30
	—	—	04	Internal Exam.	:	20

Rationale:

The subject is being introduced to produce more opportunity to practice for development of writing and oral skill both in English and Hindi language to be a good and effective communicator.

While designing the curriculum it has also been thought to promote certain student centered activities complementary to the language and communication skill. The body and sign language is also an effective method of communication and should therefore be learnt to ensure the generation of self confidence and overall personality development of the student.

S.No.	Group	Topic	Periods
1	A	ENGLISH	20
2	B	HINDI	20
3	C	SELF & SOCIETY ORIENTED ACTIVITY	10
			50

GROUP - A [ENGLISH]

S.No.	Topic	20 Periods
01	Practice on Debate, Group Discussion, Elocution and Public Speech.	
02	Practice on different role playing with emphasis on dress, behaviour, manner, personality.	
03	Practice on Letter/ Application Writing, Report Writing & Tender Notice.	

CONTENTS:

TOPIC 01 - Practice on Debate, Group Discussion, Elocution and Public Speech.:

The institute is free to undertake any topic that is current and relevant to the present need of individual, society, industrial growth, environment related to health, hygiene and sanitation, technological development and social problems etc. and a record of at least six topics is necessarily to be prepared for sessional examination.

TOPIC 02 - Practice on different role playing with emphasis on dress, behaviour, manner, personality:

- 02.01 As an executive/ supervisor
- 02.02 As an office secretary
- 02.03 As an interviewer
- 02.04 As an interviewee
- 02.05 As an office assistant
- 02.06 As a front desk operator
- 02.07 While going to a formal party

TOPIC 03 - Practice on Letter/ Application Writing, Report Writing & Tender Notice:

- 03.01 Letter/ Application writing
- 03.02 Report writing
- 03.03 Tender notice and advertisement

NOTE: Contents of the above topics are same as covered in theory papers and a record of at least two topics has to be necessarily prepared from each subtopic for sessional records.

GROUP - B [HINDI]

S.No.	Topic	20 Periods
01	तौर-तरीके एवं आधारभूत शिष्टाचार	
02	शारीरिक भाषा	
03	जानकारी(डेस्क) कार्यालय का अग्रभाग	
04	सामूहिक परिचर्चा, वाद-विवाद वक्तृता अथवा वक्तव्य	
05	पत्र/ आवेदन लेखन, प्रतिवेदन लेखन, निविदा सूचना एवं विज्ञापन	
06	अन्तर्वीक्षा के समय आचरण	

CONTENTS:

पाठ्य 01 – तौर-तरीके एवं आधारभूत शिष्टाचार:

पाठ्य 02 – शारीरिक भाषा:

- शारीरिक भाव भंगिमा द्वारा सम्प्रेषण
- अतिविहित संकेत
- मुखाकृति द्वारा सम्प्रेषण
- पोशाक तथा प्रशासन द्वारा सम्प्रेषण

पाठ्य 03 – जानकारी(डेस्क) कार्यालय का अग्रभाग:

- टेलीफोन वार्तालाप(श्रुष्टा/दृष्टा)
- कैसेट द्वारा अभ्यास

पाठ्य 04 – सामूहिक परिचर्चा, वाद-विवाद वक्तृता अथवा वक्तव्य:

उपर्युक्त पाठ्य के विषयों का चयन संस्थानों द्वारा ही निश्चित किये जायेंगे। विषय की प्रासंगिकता वर्तमान परिपेक्ष्य में हो तथा सामाजिक, औद्योगिक, स्वास्थ्य, वातावरण इत्यादि क्षेत्रों से सम्बन्धित हो।

पाठ्य 05 – पत्र/आवेदन लेखन, प्रतिवेदन लेखन, निविदा सूचना एवं विज्ञापन:

सात्रिक परीक्षा हेतु प्रत्येक से सम्बन्धित दो विषयों पर रिकार्ड तैयार करना आवश्यक होगा।

पाठ्य 06 – अन्तर्वीक्षा के समय आचरण:

GROUP - C [SELF & SOCIETY ORIENTED ACTIVITY]**Self Promotion and society oriented activity:****10 Periods**

- Library study - Assessment will be made on the basis of exposition through debate
- To create awareness among rural mass about rural technology, sanitation, health and hygiene, drinking water etc.
- Participation in cultural activity
- Any other activity taken up by the institution - related to environment